Lecture Quiz 3 Solutions - Monday, February 4, 2008

1. Solve the DE: $\frac{dy}{dt} + \frac{4}{t}y = \frac{1}{t^4}$. [8 points total]

Common sense will dictate that integrating factor should be used...

Find the integrating factor: $\mu(t) = e^{\int p(t)dt} = e^{\int \frac{4}{t}dt} = e^{4\ln|t|} = t^4$ 2 points for this (they <u>must</u> show an absolute value) Use the integrating factor: $t^4\left(y'+\frac{4}{t}y\right)=1$ **1** point for this Modify the left side: $\frac{d}{dt}(t^4y) = 1 \Longrightarrow d(t^4y) = dt$ **1** point for this Integrate both sides: $\int d(t^4 y) = \int dt \Rightarrow t^4 y = t + C$ 2 points for this (they must have a constant) Solve for y: $y = \frac{1}{t^3} + \frac{C}{t^4}$ or $y = \frac{t+C}{t^4}$ 2 points for this Is $y''' - 3ty' + 2y^2 = t^3$ linear or nonlinear? What is the order of the DE? [2 points total] 2. This equation is nonlinear. 1 point Its order is three (or third). 1 point A radioactive substance has a half-life of 3 days. We begin with 12 g of this substance. How much of

3. A radioactive substance has a half-life of 3 days. We begin with 12 g of this substance. How much of the substance remains after 1 week (*nearest tenth*)? [8 points total]

The standard growth equation is $y = y_0 e^{kt}$	2 points
	(in case full development is shown)
We start with 12 g, so $y = 12e^{kt}$	1 point
Use half-life to find k: $6 = 12e^{3k} \Rightarrow e^{3k} = \frac{1}{2} \Rightarrow k = \frac{-\ln 2}{3}$,	
so $k \approx -0.23104906018664843647241070715273$	3 points
Equation becomes $y = 12e^{-0.231t} \dots$ Find <i>y</i> when $t = 7$: $y = 12e^{-0.231(7)}$	
$\approx 2.3811015779522992121275584589085$ or 2.4 g	2 points

As specified in the syllabus, two points are deducted for missing name, ID, or recitation section number. Thus, these last two points for required information make this a 20 point quiz.

Lecture Quiz 3 Solutions - Monday, February 4, 2008

1. Solve the DE: $\frac{dy}{dt} + \frac{2}{t}y = \frac{1}{t^2}$. [8 points total]

Common sense will dictate that integrating factor should be used...

Find the integrating factor: $\mu(t) = e^{\int p(t)dt} = e^{\int_{t}^{2} dt} = e^{2\ln|t|} = t^{2}$ Use the integrating factor: $t^{2}\left(y' + \frac{2}{t}y\right) = 1$ Modify the left side: $\frac{d}{dt}(t^{2}y) = 1 \Rightarrow d(t^{2}y) = dt$ I point for this Integrate both sides: $\int d(t^{2}y) = \int dt \Rightarrow t^{2}y = t + C$ Solve for y: $y = \frac{1}{t} + \frac{C}{t^{2}}$ or $y = \frac{t+C}{t^{2}}$ 2 points for this 1 point for this 2 points for this 2 point for this

This equation is linear.	1 point
Its order is two (or second).	1 point

3. A radioactive substance has a half-life of 5 days. We begin with 16 g of this substance. How much of the substance remains after 2 weeks (*nearest tenth*)? [8 points total]

The standard growth equation is $y = y_0 e^{kt}$	2 points
	(in case full development is shown)
We start with 16 g, so $y = 16e^{kt}$	1 point
Use half-life to find k: $8 = 16e^{5k} \Rightarrow e^{5k} = \frac{1}{2} \Rightarrow k = \frac{-\ln 2}{5}$,	
so $k \approx -0.13862943611198906188344642429164$	3 points
Equation becomes $y = 16e^{-0.1386t} \dots$	
Find y when $t = 14$: $y = 16e^{-0.1386(14)}$	
$\approx 2.2973967099940700135972538935559$ or 2.3 g	2 points

As specified in the syllabus, two points are deducted for missing name, ID, or recitation section number. Thus, these last two points for required information make this a 20 point quiz.

Lecture Quiz 3 Solutions – Monday, February 4, 2008

1. Solve the DE:
$$\frac{dy}{dt} + \frac{3}{t}y = \frac{1}{t^3}$$
. [8 points total]

Common sense will dictate that integrating factor should be used...

Find the integrating factor: $\mu(t) = e^{\int p(t)dt} = e^{\int \frac{3}{t}dt} = e^{3\ln|t|} = t^3$ 2 points for this Use the integrating factor: $t^3\left(y'+\frac{3}{t}y\right)=1$ **1** point for this Modify the left side: $\frac{d}{dt}(t^3y) = 1 \Rightarrow d(t^3y) = dt$ **1** point for this Integrate both sides: $\int d(t^3 y) = \int dt \Rightarrow t^3 y = t + C$ 2 points for this (they must have a constant) Solve for y: $y = \frac{1}{t^2} + \frac{C}{t^3}$ or $y = \frac{t+C}{t^3}$ 2 point for this Is $y'' - 3ty' + 2y = t^3$ linear or nonlinear? What is the order of the DE? [2 points total] 2.

This equation is linear.	1 point
Its order is two (or second).	1 point

A radioactive substance has a half-life of 2 days. We begin with 12 g of this substance. How much of 3. the substance remains after 1 week (*nearest tenth*)? [8 points total]

The standard growth equation is $y = y_0 e^{kt}$	2 points
	(in case full development is shown)
We start with 12 g, so $y = 12e^{kt}$	1 point
Use half-life to find k: $6 = 12e^{2k} \Rightarrow e^{2k} = \frac{1}{2} \Rightarrow k = \frac{-\ln 2}{2}$,	
so $k \approx -0.34657359027997265470861606072909$	3 points
Equation becomes $y = 12e^{-0.3466t}$ Find y when $t = 7$: $y = 12e^{-0.3466(7)}$	
≈ 1.0606601717798212866012665431573 or 1.1 g	2 points

As specified in the syllabus, two points are deducted for missing name, ID, or recitation section number. Thus, these last two points for required information make this a 20 point quiz.

Brown version

(they <u>must</u> show an absolute value)

Lecture Quiz 3 Solutions - Monday, February 4, 2008

1. Solve the DE:
$$\frac{dy}{dt} + \frac{5}{t}y = \frac{1}{t^5}$$
. [8 points total]

Common sense will dictate that integrating factor should be used...

Find the integrating factor: $\mu(t) = e^{\int p(t)dt} = e^{\int \frac{5}{t}dt} = e^{5\ln|t|} = t^5$ Use the integrating factor: $t^5\left(y' + \frac{5}{t}y\right) = 1$ Modify the left side: $\frac{d}{dt}(t^5y) = 1 \Rightarrow d(t^5y) = dt$ Integrate both sides: $\int d(t^5y) = \int dt \Rightarrow t^5y = t + C$ Solve for y: $y = \frac{1}{t^4} + \frac{C}{t^5}$ or $y = \frac{t+C}{t^5}$ 2 points for this (*they must show an absolute value*) 2 points for this (*they must have a constant*) 2 point for this 2 point for this 2 point for this 2 point for this

This equation is nonlinear.	1 point
Its order is two (or second).	1 point

3. A radioactive substance has a half-life of 4 days. We begin with 16 g of this substance. How much of the substance remains after 2 weeks (*nearest tenth*)? [8 points total]

The standard growth equation is $y = y_0 e^{kt}$	2 points
	(in case full development is shown)
We start with 16 g, so $y = 16e^{kt}$	1 point
Use half-life to find k: $8 = 16e^{4k} \Rightarrow e^{4k} = \frac{1}{2} \Rightarrow k = \frac{-\ln 2}{4}$,	
So $k \approx -0.17328679513998632735430803036454$	3 points
Equation becomes $y = 16e^{-0.1733t}$ Find y when $t = 14$: $y = 16e^{-0.1733(14)}$	
≈ 1.4142135623730950488016887242097 or 1.4 g	2 points

As specified in the syllabus, two points are deducted for missing name, ID, or recitation section number. Thus, these last two points for required information make this a 20 point quiz.